High-Entropy Alloy for Hydrogen Production

Source: BL

Researchers have developed a high-entropy alloy (HEA) catalyst for water electrolysis, enhancing hydrogen production while reducing dependence on costly materials like platinum for clean energy generation.

- Alloys & High-Entropy Alloys (HEAs): Alloys are metallic substances composed of 2 or more elements, while HEAs are advanced metallic alloys materials with 5 or more elements mixed in equal or similar proportions.
 - HEA catalyst consists of platinum, palladium, cobalt, nickel, and manganese.
- HEAs possess high strength, corrosion, and wear resistance, ensuring durability.

HEA Role in Electrolysis:

- In electrolysis, a catalyst (like platinum) is used which reduces the minimum amount of energy required to start a chemical reaction (activation energy), accelerating water splitting into hydrogen and oxygen.
- The HEA catalyst reduces platinum use by 7 times improving efficiency over pure platinum, and remains stable for 100+ hours in alkaline seawater, enabling cost-effective hydrogen production.

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