

Li-ion Battery Recycling Technology to Boost Circular Economy

Why in News?

The **Ministry of Electronics and Information Technology (MeitY)** in India has taken a significant step towards promoting a <u>circular economy</u> by transferring a <u>cost-effective Li-ion battery</u> recycling technology to nine recycling industries and start-ups.

- The technology was developed under the "Centre of Excellence on E-waste management" at the Centre for Materials for Electronics Technology (C-MET), Hyderabad, in collaboration with the Government of Telangana and industry partner M/s Greenko Energies Pvt. Ltd., Hyderabad.
- This initiative is part of the <u>Mission Lifestyle for the Environment (LiFE)</u> under the "Promote circularity campaign."

What is the Recently Invented Recycling Technology?

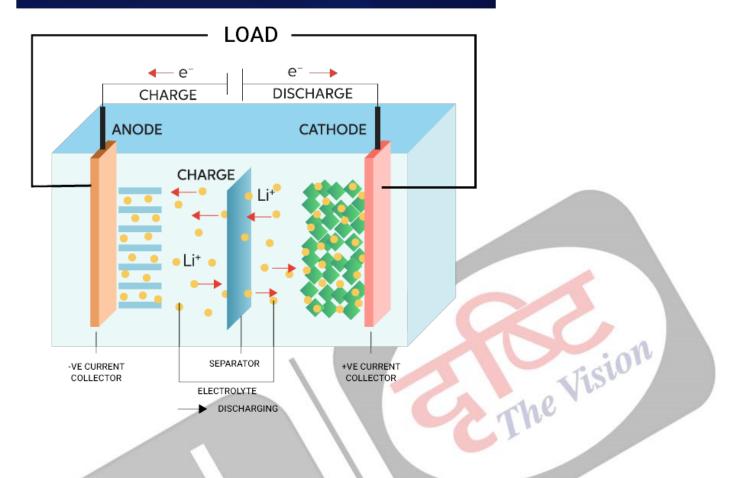
- The recycling technology for Li-ion batteries is designed to efficiently process and recover valuable materials from discarded batteries.
- The process begins by soaking the batteries in a solution to extract the valuable metals.
 - The solution aids in the separation and extraction of metals such as Lithium (Li), Cobalt (Co), Manganese (Mn), and Nickel (Ni), enabling the recovery of over 95% of their contents in the form of corresponding oxides/carbonates with a purity of approximately 98%.
- These metals are then transformed into their pure forms, ready to be reused in making new batteries or other useful applications.
- This technology ensures that over 95% of these valuable metals are recovered from batteries.
- By recycling the batteries, we can reduce the need for mining new resources and contribute to a more sustainable environment.
- The recycling technology for Li-ion batteries plays a crucial role in promoting a circular economy.

What is a Li-ion Battery?

About:

- A lithium-ion (Li-ion) battery is a type of **rechargeable battery.**
- Li-ion batteries use an intercalated (Intercalation is the reversible inclusion or insertion of a
 molecule into materials with layered structures) <u>lithium</u> compound as one electrode
 material, compared to the metallic lithium used in a non-rechargeable lithium
 battery.
- The battery consists of an **electrolyte, which allows for ionic movement**, and the two electrodes are the constituent components of a lithium-ion battery cell.
- Lithium ions move from the negative electrode to the positive electrode during discharge and back when charging.

COMPONENTS OF LITHIUM-ION BATTERY



Applications:

- Electronic gadgets, Tele-communication, Aerospace, Industrial applications.
- Lithium-ion battery technology has made it the favourite power source for electric and hybrid electric vehicles.

Disadvantages of Li-ion Batteries:

- Long charging times.
- Safety issues as instances of batteries catching fires have been there.
- Expensive to manufacture.
- While the Li-ion batteries are seen as sufficiently efficient for applications such as phones and laptops, in case of EVs, these cells still lack the range that would make them a viable alternative to internal combustion engines.

What is Lithium?

About:

• Lithium (Li), sometimes also referred as **'White gold'** due to its high demand for rechargeable batteries, is a soft and silvery-white metal.

• Extraction:

Lithium can be extracted in different ways, depending on the type of the deposit —
generally either through solar evaporation of large brine pools, or from hard-rock
extraction of the ore.

Uses:

• Lithium is an important component of **electrochemical cells** used in batteries of EVs, Laptops, Mobiles etc.

- It is also used in thermonuclear reactions.
- **It is used to make alloys** with aluminium and magnesium, improving their strength and making them lighter.
 - Magnesium-lithium alloy for armour plating.
 - Aluminum-lithium alloys in aircraft, bicycle frames and high-speed trains.
- Major Global Lithium Reserves:
 - Chile > Australia > Argentina are top countries with Li reserves.
 - Lithium Triangle: Chile, Argentina, Bolivia.
- Lithium Reserves in India:
 - Preliminary survey showed estimated lithium reserves of 14,100 tonnes in a small patch of land surveyed in Southern Karnataka's Mandya district.
 - Other potential sites:
 - Mica belts in Rajasthan, Bihar, Andhra Pradesh.
 - Pegmatite belts in Odisha and Chhattisgarh.
 - Rann of Kutch in Gujrat.

UPSC Civil Services Examination, Previous Year Question (PYQ)

- Q. Which one of the following pairs of metals constitutes the lightest metal and the heaviest metal, respectively? (2008)
- (a) Lithium and mercury
- **(b)** Lithium and osmium
- (c) Aluminium and osmium
- (d) Aluminium and mercury

Ans: (b)

Exp:

- Light metals are metals of low atomic weight while heavier elements generally have high atomic weight.
- Osmium is a hard metallic element which has the greatest density of all known elements. Osmium has an atomic weight of 190.2 u and its atomic number is 76.
- Lithium having an atomic number 3 and atomic weight of 6.941u is the lightest known metal.
- Therefore, option (b) is the correct answer.

Source: PIB

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