

Removal of Toxic Chromium from Wastewater

Source: PIB

Recently, researchers at the Institute of Nano Science and Technology (INST) Mohali, an autonomous institute of **Department of Science and Technology** have developed an innovative, method for **removing toxic <u>chromium</u> from industrial wastewater** by utilising **sunlight as a catalyst** in combination with microfluidic technology(manipulation and control of fluids at a very small scale).

- Hexavalent chromium (Cr(VI)) is highly toxic, with World Health Organization (WHO)
 recommended limits of 0.05 mg/L and 5 mg/L for hexavalent and trivalent forms in
 drinking water, necessitating the reduction of hexavalent to trivalent chromium.
 - Trivalent chrome is believed to be less readily absorbed by the human body compared to hexavalent chrome making its removal critical.
- Traditional methods like ion exchange, adsorption, and bacterial reduction are costly and often inefficient.
 - INST researchers have used sunlight as a catalyst in combination with microfluidic technology and TiO₂ nanoparticles to convert Cr(VI) into a less harmful trivalent form.
 The process has shown a 95% efficiency in degradation.
- Nanotechnology is the science of manipulating matter at the atomic or molecular scale, typically between 1 and 100 nanometers.
 - It has a wide range of applications across various fields, including: Biomedicine, Electronics,
 Removal of pollutants and toxins from water and soil, cosmetics and Food Science.

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