

Few-Electron Bubbles in Superfluid Helium Gas

Why in News

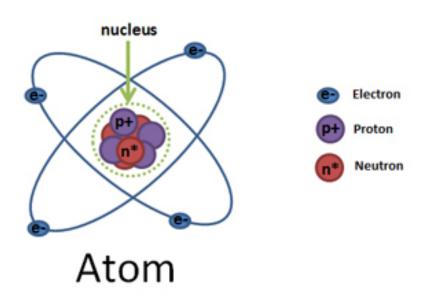
Recently, Scientists at the **Indian Institute of Science (IISc)**, Banglore for the first time discovered **two species of Few-Electron Bubbles (FEBs) in Superfluid <u>Helium Gas.</u>**

Helium

- It is a chemical element with the symbol He and atomic number 2. The British chemist Sir
 William Ramsay discovered the existence of helium on Earth in 1895.
- It is a colourless, odourless, tasteless, non-toxic, inert, monatomic gas, the first in the noble gas group in the periodic table.
- Its boiling point is the lowest among all the elements.

Electrons

- Matter is made up of atoms, which are the basic units of chemical elements such as hydrogen, helium or oxygen.
- Atoms are made up of three particles: Protons, Neutrons and Electrons.
- Hence, electrons are the subatomic particles that orbit the nucleus of an atom. They are generally negative in charge and are much smaller than the nucleus of the atom.



Key Points

• Electron Bubble:

- An electron bubble is the empty space created around a free electron in a cryogenic gas or liquid, such as neon or helium. They are typically very small, about 2 nm in diameter at atmospheric pressure.
- An electron injected into a superfluid form of helium creates a Single Electron Bubble
 (SEB) a cavity that is free of helium atoms and contains only the electron. The shape of
 the bubble depends on the energy state of the electron.
 - For instance, the bubble is spherical when the electron is in the ground state (i.e. state of lowest energy). There are also **multiple electron bubbles** that contain **thousands of electrons.**
 - Superfluidity is the frictionless flow and other exotic behaviour observed in liquid helium at temperatures near absolute zero (-273.15 °C), and similar frictionless behaviour of electrons in a superconducting solid. In each case the unusual behaviour arises from quantum mechanical effects.

Few-Electron Bubbles:

- FEBs, on the other hand, are nanometre-sized cavities in liquid helium containing
 just a handful of free electrons. The number, state, and interactions between free
 electrons dictate the physical and chemical properties of materials.
 - FEBs form an interesting system that has both **electron-electron interaction and electron-surface interaction.**
 - FEBs were found to be **stable for at least 15 milliseconds** (quantum changes typically happen at much shorter time scales) which would enable researchers to trap and study them.

Significance:

Study Properties:

• FEBs can serve as a useful model to study how the energy states of electrons and interactions between them in a material influence its properties.

Decipher Phenomenons:

- There are several phenomena that FEBs can help scientists decipher, such as:
 - Turbulent flows in superfluids and viscous fluids, or the flow of heat in superfluid helium.
 - Just like how current flows without resistance in superconducting materials at very low temperatures, superfluid helium also conducts heat efficiently at very low temperatures.

Source: DTE

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