National Green Hydrogen Mission

Source: PIB

Why in News?

Recently, the <u>Ministry of New & Renewable Energy (MNRE</u>) has increased the yearly allocation of <u>Green</u> <u>Ammonia</u> for the <u>fertiliser sector</u> from 550,000 to 750,000 tonnes to meet rising demand, enhancing support for <u>Green Hydrogen in India</u>.

What is the National Green Hydrogen Mission (NGHM)?

- India launched the National Green Hydrogen Mission (NGHM) in January 2023.
- The Ministry of New & Renewable Energy (MNRE) is implementing the NGHM with a target to achieve a production capacity of 5 million tonnes per annum of Green Hydrogen in the country by the year 2030.
 - The <u>Strategic Interventions for Green Hydrogen Transition (SIGHT)</u> programme, under NGHM, provides incentives for the manufacturing of **electrolysers and the production** of green ammonia.
- Under NGHM a dedicated portal was launched to provide information on the mission and steps for developing the green hydrogen ecosystem in India.
- India has also released scheme guidelines for the use of Green Hydrogen in steel, transport, and shipping sectors.
- The Department of Science and Technology has initiated <u>Hydrogen Valley Innovation Clusters</u> to foster innovation and promote the green hydrogen ecosystem in India.

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NATIONAL GREEN HYDROGEN MISSION

HYDROGEN H2

NODAL MINISTRY

OBJECTIVE

Ministry of New and Renewable Energy

COMPONENTS OF NGHM

Strategic Interventions for Green Hydrogen Transition Programme (SIGHT)

□ Important Description Partnership (SHIP) (PPP for R&D)

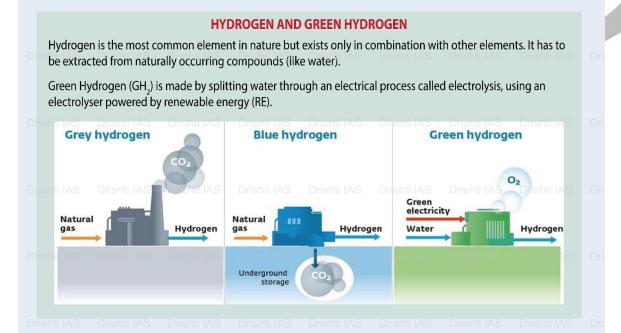
GH, is not commercially viable at present; current cost in India is around ₹350-400/kg. The National Hydrogen Energy Mission aims to bring it down under ₹100/kg.

- Decarbonise energy/industrial/mobility sector
- Develop indigenous manufacturing capacities
- Create export opportunities for GH, and its derivative

Expected Outcomes by 2030

ion

- Atleast 5MMT GH, annual production
- Rs 1 lakh crore fossil fuel import savings
- ♦ 6 lakh jobs
- 50MMT CO, annual emissions averted
- ◆ ₹ 8 lakh crore investment



Other Initiatives Related to Renewable Energy

- Jawaharlal Nehru National Solar Mission (INNSM).
- International Solar Alliance.
- PM- KUSUM.
- National Wind-Solar Hybrid Policy.
- Rooftop Solar Scheme

What is Green Ammonia?

- About:
 - <u>Ammonia</u> is a chemical that is used mainly in the **manufacture of** <u>nitrogenous fertilisers</u>, like urea and ammonium nitrate, but can be put to other uses too, such as to run engines.
 - Green ammonia production is where the process of making ammonia is 100% renewable and carbon-free.
- Method of Production:
 - It is produced by **using hydrogen from water electrolysis** and nitrogen separated from the air. These are then fed into the <u>Haber process</u>, all powered by sustainable electricity.
 - Green ammonia production makes use of renewable energy sources such as hydroelectric, solar power or wind turbines.
 - In the Haber process, hydrogen and nitrogen are reacted together at high temperatures and pressures to produce ammonia (NH3).
- Uses:
 - Energy Storage: Ammonia is easily stored in bulk as a liquid at modest pressures (10-15 bar) or refrigerated to -33°C. This makes it an ideal chemical store for renewable energy.
 Zero-carbon Fuel: Ammonia can be burnt in an engine or used in a fuel cell to produce
 - electricity. When used, ammonia's only by-products are water and nitrogen.
 - **Marine Industry:** The <u>maritime industry</u> is likely to be an early adopter, replacing the use of fuel oil in marine engines.
- Significance:
 - Green ammonia is intended to be used in the production of <u>carbon-neutral</u> fertiliser products, decarbonizing the food value chain, and also has potential as a future climateneutral shipping fuel.
 - Green ammonia is crucial to tackling the **existential challenges of producing** enough food to feed a growing global population and generating CO2-free energy.

UPSC Civil Services Examination, Previous Year Question (PYQ)

<u>Prelims:</u>

Q. Hydrogen fuel cell vehicles produce one of the following as "exhaust" (2010)

- (a) NH3
- **(b)** CH4
- **(c)** H2O
- (d) H2O2
- Ans: (c)

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